

Enantiodivergent Synthesis of Both Enantiomers of Gypsy Moth Pheromone Disparlure

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MeO N OMe
$$R = TBDMS$$
, Ts OBn

OR $C_{10}H_{21}$

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OBn

Enantiodivergent synthesis of both (-)- and (+)-disparlure, a bioactive pheromone, possessing a *cis*-epoxide has been accomplished. The key step involves the cross metathesis of a chiral homoallylic alcohol derived from L-(+)-tartaric acid.

Disparlure (*cis*-7,8-epoxy-2-methyloctadecane) **1** is the sex pheromone produced by the female gypsy moth, *Porthetria dispar L.*, a widespread forest pest, which causes severe forest losses during outbreaks. I Iwaki et al. established the configuration of **1** by synthesis and also observed that natural (+)-**1** is relatively more active than the (-)-enantiomer. It has been recently shown that the two pheromone binding proteins from the gypsy moth bind differently for both enantiomers of disparlure. Because of the different bioactivity associated with both antipodes, disparlure has been a synthetic target for a long time. Of the many enantioselective syntheses disclosed for **1**, most of the approaches utilize chiral pool starting materials, fair chiral stannanes, caymmetric chloroallylboration, be sides Sharpless asymmetric epoxidation, dihydroxylation, and enzymatic procedures.

Our interest in the synthesis of bioactive natural products from chiral pool tartaric acid resulted in the synthesis of a number of pheromones.⁶ Herein, we disclose the synthesis of both enantiomers of **1**, utilizing cross metathesis of a homoallylic alcohol, derived from L-(+)-tartaric acid. As shown in retrosynthesis (Scheme 1), synthesis of both enantiomers of disparlure is envisaged from the diol **8**, by stereoselective cyclization of **8** to either enantiomer of the epoxide. Synthesis of diol **8** was anticipated from cross metathesis of homoallylic alcohol **7** and 4-methyl-1-pentene. Allylation of benzyloxyaldehyde **6** was identified for the synthesis of homoallylic alcohol **7**. Synthesis of aldehydes similar to **6** from tartaric acid has been established in our laboratory.⁷

The synthetic sequence commenced with the addition of *n*-decylmagnesium bromide to the bis-Weinreb amide 2⁸ derived from L-(+)-tartaric acid, affording the diketone 3 in 92% yield. Under conditions that were optimized for the reduction of similar ketones, reduction of diketone 3 with K-selectride furnished the 1,4-diol 4 as the single diastereomer. Protection of 1,4-diol 4 as the dibenzyl ether and further deprotection of the acetonide resulted in 1,2-diol 5 in 89% yield. Treatment of diol 5 with Pb(OAc)₄ produced the aldehyde 6, which on allylation with allyltributyltin under Keck allylation¹⁰ conditions afforded the homoallylic alcohol 7 (Scheme 2), in 79% yield in two steps. Homoallylic alcohol 7 served as the precursor for the synthesis of both enantiomers of 1.The synthesis of (-)-disparlure (Scheme 3) was accomplished as follows. Homoallylic alcohol 7 was converted to the tosylate 9, which underwent facile cross

(5) For recent synthesis of disparlure, see: (a) Koumbis, A. E.; Chronopoulos, D. D. Tetrahedron Lett. 2005, 46, 4353. (b) Fukusaki, E.; Satoda, S.; Senda, S.; Omata, T. J. Biosci. Bioeng. 1999, 87, 103. (c) Marshall, J. A.; Jablonowski, J. A.; Jiang, H. J. Org. Chem. 1999, 64, 2152. (d) Hu, S.; Jayaraman, S.; Oehlschlager, A. C. J. Org. Chem. 1999, 64, 2152. (d) Hu, S.; Standaman, S.; Oehlschlager, A. C. J. Org. Chem. 1999, 64, 2152. (d) Hu, S.; Jayaraman, S.; Oehlschlager, A. C. J. Org. Chem. 1999, 64, 2152. (d) Hu, S.; Jayaraman, S.; Oehlschlager, A. C. J. Org. Chem. 1999, 68, 2152. (d) Hu, S.; Sayamatry, 1997, 8, 375. (f) Li, L. H.; Wang, D.; Chan, T. H. Tetrahedron: Asymmetry 1997, 8, 375. (f) Li, L. H.; Wang, D.; Chan, T. H. Tetrahedron Lett. 1997, 38, 101. (g) Curci, R.; D'Accolti, L.; Fiorentino, M.; Rosa, A. Tetrahedron Lett. 1995, 36, 5831. (h) Sinha-Bagchi, A.; Sinha, S. C.; Keinan, E. Tetrahedron: Asymmetry 1995, 60, 2889. (i) Paolucci, C.; Mazzini, C.; Fava, A. J. Org. Chem. 1995, 60, 169. (j) Ko, S. Y. Tetrahedron Lett. 1994, 35, 3601. (k) Tsuboi, S.; Furutani, H.; Ansari, M. H.; Sakai, T.; Utaka, M.; Takeda, A. J. Org. Chem. 1993, 58, 486. (l) Brevet, J.-L.; Mori, K. Synthesis 1992, 1007. (m) Keinan, E.; Sinha, S. C.; Sinha-Bagchi, A.; Wang, Z.-M.; Zhang, X.-L.; Sharpless, K. B. Tetrahedron Lett. 1992, 33, 6411.

(6) (a) Prasad, K. R.; Anbarasan, P. Tetrahedron: Asymmetry 2005, 16, 3951. (b) Prasad, K. R.; Anbarasan, P. Tetrahedron Lett. 2006, 47, 1433. (c) Prasad, K. R.; Anbarasan, P. Synlett 2006, 2087. (d) Prasad, K. R.; Anbarasan, P. Tetrahedron: Asymmetry 2006, 17, 1146. (e) Prasad, K. R.; Anbarasan, P. Tetrahedron 2006, 62, 8303. (f) Prasad, K. R.; Chandrakumar, A.; Anbarasan, P. Tetrahedron: Asymmetry 2006, 17, 1979. Synthesis of styryl lactones: (g) Prasad, K. R.; Gholap, S. L. Synlett 2005, 2260. (h) Prasad, K. R.; Gholap, S. L. J. Org. Chem. 2006, 71, 3643.

(7) Prasad, K. R.; Chandrakumar, A. Tetrahedron: Asymmetry 2005, 16, 1897.

(8) (a) Nugiel, D. A.; Jakobs, K.; Worley, T.; Patel, M.; Kaltenbach, R. F., III; Meyer, D. T.; Jadhav, P. K.; De Lucca, G. V.; Smyser, T. E.; Klabe, R. M.; Bacheler, L. T.; Rayner, M. M.; Seitz, S. P. *J. Med. Chem.* **1996**, *39*, 2156. (b) McNulty, J.; Grunner, V.; Mao, J. *Tetrahedron Lett.* **2001**, *42*, 5609.

(9) Formation of the other possible diastereomers was not observed within detectable limits by NMR. For a systematic study on the factors governing the diastereoselectivity in the reduction and Grignard reagent addition to similar C_2 -symmetric diketones derived from tartaric acid, see: Prasad, K. R.; Chandrakumar, A. *Synthesis* **2006**, 2159.

^{*} Corresponding author. Fax: +918023600529.

⁽¹⁾ Bierl, B. A.; Beroza, M.; Collier, C. W. Science 1970, 170, 87.

⁽²⁾ Iwaki, S.; Marumo, S.; Saito, T.; Yamada, M.; Katagiri, K. J. Am. Chem. Soc. 1974, 96, 7842.

⁽³⁾ Plettner, E.; Lazar, J.; Prestwich, E. G.; Prestwich, G. D. *Biochemistry* **2000**, *39*, 8953.

⁽⁴⁾ For a review on the synthesis of insect pheromones, see: (a) Mori, K. Curr. Org. Synth. 2004, 1, 11. (b) Mori, K. Acc. Chem. Res. 2000, 33, 102.

SCHEME 1. Retrosynthesis of Disparlure

SCHEME 2. Stereoselective Synthesis of Homoallylic Alcohol 7

metathesis¹¹ with 4-methyl-1-pentene in presence of 5 mol % of Grubbs second generation catalyst to afford the alkene 10^{12} in 92% yield. Hydrogenation of the alkene 10 with Pd/C led to the formation of hydroxytosylate 11,¹³ which on treatment with K_2CO_3 furnished (-)-1 in 89% yield: $[\alpha]_D - 1.0$ (c 1.7, CCl₄); lit.^{5c} $[\alpha]_D - 0.9$ (c 0.21, CCl₄); the spectral data of which are identical to those reported in literature.^{5c}For the synthesis of (+)-1 (Scheme 4), the alcohol group in homoallylic alcohol 7 was protected as the *tert*-butyldimethylsilyl ether under standard conditions to yield 12 in 98% yield. Cross metathesis of 12 with 4-methyl-1-pentene resulted in alkene 13,¹⁴ which on hydrogenation with Pd/C afforded the alcohol 14 in 81% yield in two steps. Reaction of 14 with TsCl and DMAP produced

SCHEME 3. Synthesis of (-)-Disparlure 1

SCHEME 4. Synthesis of (+)-Disparlure 1

the tosylate **15** in 90% yield. Treatment of **15** with TBAF furnished (+)-**1** in 89% yield: $[\alpha]_D +1.0$ (c 1.5, CCl₄); lit. ^{5c} $[\alpha]_D +0.9$ (c 1.1, CCl₄). In summary, a facile enantiodivergent synthesis of the bioactive pheromone, disparlure, was achieved. Homoallylic alcohol derived from chiral pool L-(+)-tartaric acid served as the key building block for the synthesis of both (–)-and (+)-disparlure, involving cross metathesis as the pivotal step.

Experimental Section

(4R,5R)-5-(Benzyloxy)pentadec-1-en-4-ol (7): To a solution of 5 (0.2 g, 0.34 mmol) in 3 mL of benzene at room temperature was added Pb(OAc)₄ (0.27 g, 0.6 mmol) under argon atmosphere. The reaction mixture was stirred for 1.5 h at the same temperature, quenched with water (0.2 mL), and stirred for 10 min at room temperature. The reaction mixture was filtered through a short pad of Celite, and the Celite pad was washed with CH₂Cl₂ (25 mL). The combined organic layers were dried over Na₂SO₄, and the solvent was evaporated under reduced pressure to yield α-benzyloxyaldehyde 6 as a colorless oil. It was subjected to the next reaction without purification.

A suspension of aldehyde 6 (obtained above) and MgBr₂•Et₂O (0.23 g, 0.9 mmol) in 3 mL of CH₂Cl₂ at −78 °C was stirred under argon atmosphere for 1 h. Allyltributyltin (0.3 mL, 0.9 mmol) was introduced dropwise over a period of 5 min at the same temperature. The reaction mixture was stirred for 2 h, poured into water (10 mL), and extracted with ether (3 × 8 mL). Combined ethereal extracts were washed with 1% aqueous NH₃ (three times) to remove tin impurities, brine, and dried over Na₂SO₄. Evaporation of solvent followed by column chromatography of the resulting residue using petroleum ether/EtOAc (96:4) as an eluent yielded 7 as a colorless oil in 79% (for two steps, 0.18 g): $[\alpha]_D$ -18.8 (c 1, CHCl₃); IR (neat) 3442, 2952, 2854, 1456, 1090, 1070, 912, 697 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.37-7.27 (m, 5H), 5.94-5.78 (m, 1H), 5.14-5.06 (m, 2H), 4.65 (d, J = 11.4 Hz, 1H), 4.50 (d, J = 11.4Hz, 1H), 3.67-3.60 (m, 1H), 3.36-3.28 (m, 1H), 2.39-2.18 (m, 3H), 1.71-1.49 (m, 2H), 1.41-1.18 (m, 16H), 0.88 (t, J=7.2

^{(10) (}a) Keck, G. E.; Boden, E. P. *Tetrahedron Lett.* **1984**, *25*, 265. Chelation-controlled addition of allyltributyltin to similar aldehydes is well documented in literature. (b) Schlessinger, R. H.; Graves, D. D. *Tetrahedron Lett.* **1987**, *28*, 4381. (c) Keck, G. E.; Andrus, M. B.; Romer, D. R. *J. Org. Chem.* **1991**, *56*, 417. (d) Murga, J.; Falomir, E.; Garcia-Fortanet, J.; Carda, M.; Marco, J. A. *Org. Lett.* **2002**, *4*, 3447. (e) Prasad, K. R.; Penchalaiah, K.; Choudhary, A.; Anbarasan, P. *Tetrahedron Lett.* **2007**, *48*, 309. For a review on Lewis acid mediated allylation with allyltin compounds, see: (f) Nishigaichi, Y.; Takuwa, A.; Naruta, Y.; Maruyama, K. *Tetrahedron* **1993**, *34*, 7395.

^{(11) (}a) Grubbs, R. H.; Trnka, T. M.; Sanford, M. S. Curr. Meth. Inorg. Chem. **2003**, *3*, 187. (b) Connon, S. J.; Blechert, S. Angew. Chem., Int. Ed. **2003**, *42*, 1900.

⁽¹²⁾ Evaluation of the ratio of E/Z regioisomers in 10 resulting from the cross metathesis was inconclusive from NMR. No further efforts were made to estimate the E/Z ratio. The stereochemistry of the double bond is of no consequence because it is saturated in the next step by hydrogenation.

⁽¹³⁾ A minor amount (6%) of product resulting from the displacement of the tosyl group is also observed.

⁽¹⁴⁾ It was difficult to separate the product alkene 13 from homoallylic alcohol 12 at this stage. However, further hydrogenation of the alkene 13 resulted in the alcohol 14, which was easily purified.

Hz, 3H); 13 C NMR (75 MHz, CDCl₃) δ 138.4 (Cq), 134.9 (CH), 128.4 (CH), 127.8 (CH), 127.7 (CH), 117.2 (CH₂), 81.4 (CH), 72.3 (CH₂), 71.9 (CH), 38.0 (CH₂), 31.9 (CH₂), 30.1 (CH₂), 29.8 (CH₂), 29.6 (CH₂), 29.5 (CH₂), 29.3 (CH₂), 25.1 (CH₂), 22.6 (CH₂), 14.0 (CH₃); HRMS for $C_{22}H_{36}O_2$ + Na calcd 355.2613; found 355.2612.

(7R,8R)-8-(Benzyloxy)-7-(p-toluenesulfonyloxy)-2-methyloctadec-4-ene (10): A mixture of 9 (50 mg, 0.1 mmol), 4-methyl-1-pentene (0.07 mL, 0.5 mmol), and Grubbs second generation catalyst (5 mg, 0.005 mmol) in 1 mL of CH₂Cl₂ was refluxed for 6 h. The solvent was removed, and the syrup thus obtained was chromatographed using petroleum ether/EtOAc (97:3) as an eluent to yield 10 in 92% (51 mg) as a colorless oil, as an E/Z mixture (the ratio of the geometrical isomers is inconclusive from ¹H NMR; since the next step is the reduction of the double bond, no efforts were made to separate the E/Z isomers and is used as such in the next step): IR (neat) 2924, 2853, 1599, 1455, 1367, 1188, 1176, 1098, 1055, 899 cm⁻¹; 1 H NMR (300 MHz, CDCl₃) δ 7.73 (d, J = 8.4 Hz, 2H, 7.37 - 7.24 (m, 7H), 5.41 - 5.27 (m, 1H), 5.18 -4.96 (m, 1H), 4.58-4.50 (m, 1H), 4.51 (s, 2H), 3.57-3.49 (m, 1H), 2.51-2.40 (m, 1H), 2.42 (s, 3H), 2.28-2.19 (m, 1H), 1.83-1.12 (m, 21H), 0.92–0.80 (m, 9H); $^{13}\mathrm{C}$ NMR (75 MHz, CDCl₃) δ 144.4 (Cq), 138.3 (Cq), 134.2 (Cq), 133.1 (CH), 129.6 (CH), 128.3 (CH), 127.9 (CH), 127.8 (CH), 127.6 (CH), 125.1 (CH), 83.0 (CH), 78.8 (CH), 72.6 (CH₂), 41.9 (CH₂), 32.7 (CH₂), 31.9 (CH₂), 29.6 (CH₂), 29.5 (CH₂), 29.3 (CH₂), 29.1 (CH₂), 28.2 (CH), 25.5 (CH₂), 22.7 (CH₂), 22.3 (CH₃), 22.2 (CH₃), 21.6 (CH₃), 14.1 (CH₃).

(7*R*,8*R*)-7-(*p*-Toluenesulfonyloxy)-2-methyloctadecan-8-ol (11): To a solution of 10 (40 mg, 0.07 mmol) in 2 mL of methanol at room temperature was added activated 10% Pd/C (20 mg). The reaction mixture was stirred for 3 h under hydrogen atmosphere (balloon) at the same temperature. After the reaction was complete (TLC), it was filtered through a short pad of Celite and the Celite pad was washed with ether (15 mL). Evaporation of solvent followed by column chromatography of the resulting residue using petroleum ether/EtOAc (94:6) yielded 11 as a colorless solid in 90% (30 mg) yield: mp 40.4–41.5 °C; $[\alpha]_D$ +12.3 (c 2.4, CHCl₃); lit.¹⁵ mp 41.0-41.5 °C; $[\alpha]_D$ -12.3 (c 2.0, CHCl₃) for the enantiomer; IR (neat) 3393, 2924, 2853, 1600, 1455, 1364, 1175, 1019, 896 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.81 (d, J = 8.4Hz, 2H), 7.34 (d, J = 8.4 Hz, 2H), 4.49 (dt, J = 6.0, 5.4 Hz, 1H), 3.50-3.68 (br m, 1H), 2.45 (s, 3H), 1.79 (d, J = 6.6 Hz, OH, exchangeable with D_2O), 1.75-0.95 (m, 27H), 0.88 (t, J = 7.2Hz, 3H), 0.83 (d, J=6.6 Hz, 6H); 13 C NMR (75 MHz, CDCl₃) δ 144.7 (Cq), 134.3 (Cq), 129.7 (CH), 127.7 (CH), 86.7 (CH), 71.8 (CH), 38.6 (CH₂), 32.9 (CH₂), 31.9 (CH₂), 30.6 (CH₂), 29.6 (CH₂), 29.5 (CH₂), 29.4 (CH₂), 29.3 (CH₂), 27.8 (CH₁), 27.1 (CH₂), 25.5 (CH₂), 25.1 (CH₂), 22.7 (CH₂), 22.5 (CH₃), 21.6 (CH₃), 14.1 (CH₃).

(-)-Disparlure (*cis*-7,8-Epoxy-2-methyloctadecane) (1): To a solution of 11 (25 mg, 0.055 mmol) in 1 mL of methanol was added potassium carbonate (12 mg, 0.08 mmol) at room temperature under inert atmosphere. The reaction mixture was stirred for 1 h at the same temperature, poured into water (4 mL), and extracted with ether. The combined ethereal layer was washed with brine and dried over Na₂SO₄. Evaporation of solvent followed by column chromatography of the resulting residue using petroleum ether/EtOAc (99:1) yielded (-)-1 as a colorless oil in 89% (14 mg): $[\alpha]_D$ -1.0 (*c* 1.7, CCl₄); lit.^{5c} $[\alpha]_D$ -0.9 (*c* 0.21, CCl₄); ¹H NMR (300 MHz,

(15) Mori, K.; Takigawa, T.; Matsui, M. Tetrahedron 1979, 35, 833.

CDCl₃) δ 2.91–2.87 (m, 2H), 1.57–1.15 (m, 27H), 0.90–0.84 (m, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 57.2 (CH), 38.9 (CH₂), 31.9 (CH₂), 29.5 (CH₂), 29.3 (CH₂), 27.9 (CH), 27.8 (CH₂), 27.3 (CH₂), 26.8 (CH₂), 26.6 (CH₂), 22.7 (CH₂), 22.6 (CH₃), 14.1 (CH₃).

(7R,8R)-7-(tert-Butyldimethylsilyloxy)-2-methyloctadecan-8-ol (14): A mixture of 12 (60 mg, 0.13 mmol), 4-methyl-1-pentene (0.09 mL, 0.67 mmol), and Grubbs second generation catalyst (6 mg, 0.0065 mmol) in 1 mL of CH₂Cl₂ was refluxed for 8 h. After the reaction was complete (indicated by TLC), it was concentrated to a syrup. Column chromatography of the syrup using petroleum ether/EtOAc (98:2) as an eluent yielded 13, admixed with a small amount of 12 which was not separable by column chromatography. It was subjected to the next reaction without further purification.

To a solution of the crude 13 obtained above in 2.5 mL of methanol at room temperature was added activated 10% Pd/C (30 mg). The reaction mixture was stirred for 3 h under hydrogen atmosphere (balloon) at the same temperature. After the reaction was complete (TLC), it was filtered through a short pad of Celite and the Celite pad was washed with ether (15 mL). Residue obtained after evaporation of solvent was purified by column chromatography using petroleum ether/EtOAc (98:2) to yield 14 as a colorless oil in 81% (45 mg): $[\alpha]_D$ -3.8 (c 3.2, CHCl₃); IR (neat) 3439, 2927, 2856, 1463, 1255, 1084, 836, 775 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 3.44-3.31 (m, 2H), 2.06 (d, J = 6.9 Hz, 1H, exchangeable with D₂O), 1.61-1.05 (m, 27H), 0.86-0.75 (m, 18H), 0.01 (s, 3H), -0.01 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 75.1 (CH), 72.6 (CH), 38.9 (CH₂), 36.1 (CH₂), 34.1 (CH₂), 33.9 (CH₂), 31.9 (CH₂), 29.7 (CH₂), 29.6 (CH₂), 29.3 (CH₂), 27.9 (CH), 27.6 (CH₂), 25.9 (CH₂), 25.8 (CH₃), 25.2 (CH₂), 22.6 (CH₂), 22.5 (CH₃), 18.1 (Cq), 14.1 (CH₃), -4.1 (CH₃), -4.6 (CH₃); HRMS for $C_{25}H_{54}O_{27}$ Si + Na calcd 415.3971; found 415.3978.

(+)-Disparlure (*cis*-7,8-Epoxy-2-methyloctadecane) (1): To a solution of **15** (30 mg, 0.05 mmol) in 1.5 mL of THF was added TBAF (0.15 mL, 0.15 mmol) at 0 °C under argon atmosphere. The reaction mixture was stirred for 10 h at room temperature. After the reaction was complete (indicated by TLC), it was poured into water (5 mL) and extracted with ether (3 × 8 mL). The combined ethereal extracts were washed with brine and dried over Na₂SO₄. Evaporation of solvent followed by column chromatography using petroleum ether/EtOAc (99:1) as an eluent yielded (+)-**1** in 89% as a colorless oil: [α]_D +1.0 (c 1.5, CCl₄); lit.^{5c} [α]_D +0.9 (c 1.1, CCl₄); ¹H NMR (300 MHz, CDCl₃) δ 2.93–2.88 (m, 2H), 1.59–1.15 (m, 27H), 0.91–0.84 (m, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 57.2 (CH), 38.9 (CH₂), 31.9 (CH₂), 29.5 (CH₂), 29.3 (CH₂), 27.9 (CH–), 27.8 (CH₂), 27.3 (CH₂), 26.8 (CH₂), 26.6 (CH₂), 22.6 (CH₂), 22.5 (CH₃), 14.1 (CH₃).

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Supporting Information Available: Experimental procedures and spectroscopic data for the compounds and copies of ¹H NMR and ¹³C NMR spectra are provided. This material is available free of charge via the Internet at http://pubs.acs.org.

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